

**ANIL NEERUKONDA
INSTITUTE OF TECHNOLOGY AND SCIENCES
(AUTONOMOUS)
ACCREDITED BY NBA & NAAC WITH 'A' GRADE**

Affiliated to Andhra University



**Academic Regulations
Curriculum &
Syllabi (First Year I & II Sem)**

**DEPARTMENT OF
ELECTRONICS & COMMUNICATION ENGINEERING**

**ANIL NEERUKONDA INSTITUTE OF
TECHNOLOGY AND SCIENCES
(AUTONOMOUS)**

VISION

ANITS envisions to emerge as a world-class technical institution whose products represent a good blend of technological excellence and the best of human values.

MISSION

To train young men and women into competent and confident engineers with excellent communicational skills, to face the challenges of future technology changes, by imparting holistic technical education using the best of infrastructure, outstanding technical and teaching expertise and an exemplary work culture, besides moulding them into good citizens.

QUALITY POLICY

ANITS is engaged in imparting quality technical education. It constantly strives towards achieving high standards of teaching, training and development of human resources by encouraging its faculty and staff to work as a team and to update their knowledge and skills continually to match the needs of industry.

Foreword

ANIL NEERUKONDA INSTITUTE OF TECHNOLOGY AND SCIENCES (ANITS) was founded by Anil Neerukonda Educational Society (ANES) in the fond memory of Anil Neerukonda, son of Dr. B R Prasad Neerukonda.

Its humble journey started in 2001 with an intake of 220 students into four undergraduate B.Tech programmes. Within 14 years of its establishment, the institute registered phenomenal growth and is accredited by NAAC with 'A' and by NBA for the second time. It is permanently affiliated to Andhra University and has achieved autonomous status in 2015. Further, the institute has been currently ranked as 4th among the private engineering colleges in Andhra Pradesh by APSCHE. It has been recognised as "Centre for Excellence" by Infosys and is accorded by Andhra University as "Centre for Research".

Today, the institute offers seven B.Tech. programmes and four M.Tech. programmes with an annual total intake about 1100 students. The institute offers amenities like separate hostels for boys and girls, indoor and outdoor games, transport covering all the major locations of Visakhapatnam and medical aid provided by Anil Neerukonda hospital and NRI Institute of Medical Sciences, another educational institution of ANES.

Apart from the State-of-the-Art laboratories, well established teaching methodology and implementation of the best practices, the wonderful co-ordination of the Management, Faculty and Parents has so far played a crucial role in shaping the future of the ANITIANS and has been the talisman of the Institute's phenomenal growth.

The success stories of our champions at several qualifying exams for the higher studies like GRE, TOEFL, CAT and GATE, the impressive track record of the placements with highest known packages in MNCs like Google, Oracle, Infosys, TCS and so on are the sweetest fruits of our efforts.

PRAGNANAM BRAHMA, the motto of ANITS, is truly practiced by all the members of ANITS family, a direct effort to serve the society, nation and the mankind as well.

Hearty welcome to ANITS family.

Prof. T.V. Hanumantha Rao
PRINCIPAL

Achievements & Highlights

- Autonomous since May 2015
- NAAC with 'A' Grade
- Accredited and reaccredited by NBA, New Delhi
- UGC recognition under 2(f) and 12(B)
- Permanent affiliation to Andhra University, Visakhapatnam
- Among top 3 most preferred colleges in A.P.
- "AAA" rating accorded by "Careers Digest 360"
- Recognized as a Research Center by Andhra University
- Selected as Skill Development Center (SDC) by Govt. of A.P.
- First institute to be accorded "Center for Excellence" by Infosys
- Ranked 3rd among the Promising Private Engineering Colleges for excellence as per Competition Success Review (CSR) magazine in the year 2016
- Recognized as "Silver Partner" of Keane India (Chennai) for the year 2007-2008
- Collaborated with "Mission (R&D)" funded by Wipro
- "On Campus Training" by IBM for the students
- Collaboration with Unisys Global Solutions India (Bangalore) for internship
- Highest package offer around 2 crores including perks – highest offer in South India
- 8 lacs to 10 lacs packages –for majority ANITIANS

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Department Profile

The Department of **Electronics and Communication Engineering** was started with B.E programme (UG) with an intake of 60 seats in the year 2001, subsequently enhanced to 90 seats in 2003, to 120 seats in 2005 and to 180 seats in the year 2014 and M.Tech programme (PG) in Communication Systems with an intake of 18 seats from 2011-2012. Department was recognized as Research Center by Andhra university for guiding Ph.D scholars from 2014-15. The Department was accredited twice by NBA. The institute attained autonomous status from the academic year 2015-2016 and accredited with 'A' grade by NAAC.

At present, the department has 28 qualified and dedicated faculty members with specializations of Wireless & Mobile Communications, EMI/EMC, Antennas, Microelectronics & VLSI Design, Radar & Microwave Engineering, Signal Processing, Image Processing & Computer Vision, Electronic Instrumentation, Digital Electronics & Communication Systems, 6 technical staff and 4 supporting staff supplements the teaching staff. The department has successfully completed one Research Promotion Scheme (RPS) project funded by AICTE. The faculty members are actively involved in research and are publishing papers in reputed national and international conferences/ journals.

The department has well equipped laboratories namely, Electronic Devices and Circuits Lab, Linear Integrated Circuits and Pulse Circuits Lab, Communications Lab, Digital ICs & Microprocessors Lab, Microwave and Antennas Lab, Digital Signal Processing Lab and Project/Research Lab. The laboratories are equipped with special hardware and software tools useful to train the students to meet the needs of industry such as Color TV Training module, Universal multi vendor development kit, latest configured computer systems, Microwave Benches(X-band), Spectrum Analyzer, Antenna and

Optical Fiber Training module, DSP trainer module, Lab VIEW software, MATLAB software, Tanner tools Software, Hyper-Lynx 3D EM super structure Designer for antenna design, XILINX and VLSI design software, etc.

All the faculty members are easily accessible to the students for advice, counseling and guidance on curricular, co-curricular and extra-curricular (NSS, Sports, etc.) activities. The department organizes annual student technical symposiums, in which students from various colleges across the country participate and exhibit their talents in events like paper presentation, poster presentation, hardware exhibition, technical quiz, and mock parliament. The department has student forums of professional national and international professional bodies like IETE, IEEE, etc.

ECE department has an excellent placement record which has been consistently above 85 % and the students are placed in reputed IT and core industries. The students constantly get admissions in IITs, NITs, IIMs, reputed Indian universities and foreign universities for higher studies.

**DEPARTMENT OF
ELECTRONICS & COMMUNICATION ENGINEERING**

VISION

To become a centre of excellence in Education and Research and produce high quality engineers in the field of Electronics and Communication Engineering to face the challenges of future technological changes.

MISSION

The Department aims to bring out competent young Electronics & Communication Engineers by achieving excellence in imparting technical skills, soft skills and the right attitude for continuous learning.

PROGRAM EDUCATIONAL OBJECTIVES (PEOs)

- PEO 1 : To prepare graduates for successful career in Electronics industry, R&D organizations and/or IT industry by providing technical competency in the field of Electronics & Communication Engineering.

- PEO 2 : To prepare graduates with good scientific and engineering proficiency to analyze and solve electronics engineering problems.

- PEO 3 : To inculcate in students professionalism, leadership qualities, communication skills and ethics needed for a successful professional career.

- PEO 4 : To provide strong fundamental knowledge in students to pursue higher education and continue professional development in core engineering and other fields.

PROGRAM OUTCOMES (POs)

At the end of the program, the students will have

- PO 1 An ability to apply knowledge of mathematics, science and engineering with adequate computer knowledge to electronics and communication engineering problems.
- PO 2 An ability to analyze complex engineering problems through the knowledge gained in core electronics engineering and interdisciplinary subjects appropriate to their degree programme.
- PO 3 An ability to design, implement and test an electronics based system.
- PO 4 An ability to design and conduct scientific and engineering experiments, as well as to analyze and interpret data.
- PO 5 An ability to use modern engineering techniques, simulation tools and skills to solve engineering problems.
- PO 6 An ability to apply reasoning in professional engineering practice to assess societal, safety, health and cultural issues;
- PO 7 An ability to understand the impact of professional engineering solutions in societal and environmental contexts;
- PO 8 An ability to develop skills for employability/ entrepreneurship and to understand professional and ethical responsibilities;
- PO 9 An ability to function effectively as an individual on multi-disciplinary tasks.
- PO 10 An ability to convey technical material through oral presentation and interaction with audience, formal written papers /reports which satisfy accepted standards for writing style;
- PO 11 An ability to succeed in university and competitive examinations to pursue higher studies;
- PO 12 An ability to recognize the need for and engage in life-long learning process.

PROGRAM SPECIFIC OUTCOME (PSOs)

- PSO 1 Competency in the application of circuit analysis and design.
- PSO 2 The ability to apply knowledge of physics/chemistry/ mathematics to electronic circuits.
- PSO 3 The ability to apply the knowledge of computer programming, analog & digital electronics, microprocessors, etc and associated software to design and test VLSI / communication systems.
- PSO 4 The ability to pursue higher studies either in India or Abroad in specializations like communication systems. VLSI, embedded systems, signal processing, image processing, RADAR & Microwave engineering, etc and also lead a successful career with professional ethics.

ACADEMIC REGULATIONS

ACADEMIC REGULATIONS FOR B.TECH PROGRAMME UNDER AUTONOMOUS STATUS

(W.E.F. THE ADMITTED BATCH OF 2015-16)

I. Admissions:

Admissions into first year of B.Tech.Programme and admissions into second year (lateral entry) of B.Tech.Programme of the Institute will be as per the norms stipulated by Andhra University & Andhra Pradesh State Council for Higher Education (APSCHE), Govt. of Andhra Pradesh. The academic regulations of Autonomous status mentioned herewith will be applicable from 2016-17 in case of Lateral Entry admissions.

II. Programmes Offered:

The following are the B.Tech. Programmes offered by the Institute.

- 1 Chemical Engineering
- 2 Civil Engineering
- 3 Computer Science & Engineering
- 4 Electrical & Electronics Engineering
- 5 Electronics & Communication Engineering
- 6 Information Technology
- 7 Mechanical Engineering

III. Structure of the B. Tech. Programme:

The programme consists of Humanities, Basic Sciences, Engineering Sciences and Technology. The complete programme is distributed over eight semesters with two semesters per academic year. Every branch of B.Tech programme will have a curriculum and syllabi for the courses recommended by the Board of Studies and approved by the Academic Council. The academic programmes of the Institute follow the credit system. The curriculum of B.Tech programme is designed to have a total of about 189 credits of which a student should acquire a minimum of 180 credits to get the degree awarded. If a student earns all the total credits, then the best 180 credits are considered to determine the final CGPA. The lateral entrants shall have a total of about 146 credits of which one should acquire a minimum of 137 credits to get the degree awarded. If a lateral student takes all the credits, then the best 137 credits are considered to determine the final CGPA.

Criteria for achieving the minimum credits:

❖ Mandatory courses

All courses mentioned in the programme excluding open electives, professional electives and MOOCS come under mandatory courses.

❖ Open Elective- A course offered by any department other than home department

The student has to choose one open elective out of the open electives offered by other departments during third year first semester or Final year first semester.

Professional Electives

The student has to register for at least (n-1) no. of professional electives (n = no. of professional electives offered by the department during the programme) as per his choice as provided in the curriculum. However, he can register for all the professional electives offered by the department.

❖ MOOCs- Massive Open Online Courses

The student is required to register for one MOOCs course any time during second year first semester to fourth year second semester. However, its grade will be accorded at the end of fourth year second semester along with the fourth year second semester courses of the programme.

For the award of the degree, the student has to secure a minimum pass grade or above in all the mandatory courses, registered open elective, registered professional electives. However, the degree will still be awarded even if the student fails / opts out of MOOCs.

IV. Duration of the Programme:

The duration of the programme is four academic years consisting of two semesters in each academic year. A student is permitted to complete the programme in a stipulated time frame of 8 consecutive academic years from the date of initial admission. Students joining the programme in the 2nd year through lateral entry scheme shall have to complete the programme in a stipulated time frame of 6 consecutive academic years from the date of initial admission.

V. Medium of Instruction:

The medium of instruction and examination is English.

VI. Minimum Instruction Days:

Each semester normally consists of a minimum of 16 weeks of instruction.

VII. Academic Calendar:

The dates of all important events, such as commencement of class work, examinations, vacations, etc., during the academic year will be specified in the Academic Calendar of the Institute, as approved by the Academic Council.

VIII. Examinations & Evaluation Process:

The performance of a student in each semester shall be evaluated subject-wise with a maximum of 100 marks each for theory and practical/drawing subjects.

(A) Theory Course:

For all lecture based theory courses, the assessment shall be for 40 marks through internal evaluation and 60 marks through external semester-end examination of three hours duration except for the subjects with 100% internal assessment in which case an internal examination will be conducted for 60 marks along with the semester-end examinations.

i) Internal evaluation:

The sessional marks shall be awarded through internal evaluation by the teachers concerned based on the continuous assessment which includes class tests, quiz, viva-voce, assignments, student regularity, two mid-examinations etc., according to a scheme notified by the department at the beginning of the semester.

Out of the 40 internal evaluation marks, 20 marks are assigned for 2 internal-mid exams, 10 marks for assignments, 5 marks for projects/ case studies /quiz/tests and 5 marks for attendance. The average of 2 internal-mid exams is considered for the 20 marks allocated.

Under any circumstances, no re-examination shall be conducted for the internal mid examinations.

ii) External evaluation:

The question paper shall be set externally and the answer scripts are valued through a double valuation system.

The average of the two valuations will be taken for the award of marks. In case, the difference of the marks obtained in the two valuations is

more than 20%, then a third examiner shall value the script. Out of the three valuations, the average of marks obtained in third valuation and the marks obtained nearer to third valuation out of first two valuations shall be considered. No revaluation for any subject/course shall be entertained as already double valuation system is in existence. However, recounting is allowed on the request of the candidate on payment of specified fee. Challenge valuation shall also be entertained on payment of specified fee.

(B) Laboratory Course:

Each student will perform about 10 to 12 experiments in each laboratory course. Laboratory course will be evaluated for 100 marks, out of which 50 marks are for external examination and 50 marks are for internal evaluation. The internal marks are awarded based on continuous assessment, record work, internal lab examination and student regularity. The external examination will be conducted by two examiners, one of them being laboratory class teacher as internal examiner (nominated by the Principal on recommendation of HOD) and an external examiner nominated by the Principal from the panel of experts recommended by the HOD.

A candidate shall be declared to have passed any theory subject/course if he secures not less than 40% in external theory examination and also a minimum of 40% of total marks of that course which assures a minimum of 'P' grade.

A candidate shall be declared to have passed any practical course if he secures not less than 50% in external laboratory examination and also a minimum of 50% of total marks of that course which assures a minimum of 'C' grade.

Only in the case of quantitative and verbal aptitude – I & II, if a candidate fails he is given an opportunity to improve to pass grade (P) irrespective of the score he gets over and above pass mark in the reexamination within one month on payment of special examination fee.

Any student appearing for the semester-end practical examination is eligible only if he submits the bonafide record certified by the laboratory class teacher and the HOD.

(C) Project Work:

The project work is evaluated for 300 marks out of which 100 through internal assessment in the IV Year I semester through continuous assessment followed by final evaluation by a committee nominated by the HOD. For the 200 marks in IV year II semester, assessment is done for 100 marks internally and for the remaining 100 marks by the committee consisting of

at least one external expert nominated by the Principal. If a student fails in the fourth year first semester project he has to appear for reassessment within one month for which he has to pay the reexamination fee.

(D) Industrial Training:

The industrial training is assessed internally for 100 marks by an internal evaluation committee constituted by the HOD.

(E) Supplementary Exam:

There will be supplementary examination for the programme such that for odd semester courses the supplementary exams will be conducted during summer vacation and for the even semester courses, the supplementary exams will be conducted during the winter vacation.

IX. Attendance Regulations:

Attendance of a student is computed by considering total number of periods conducted in all courses as the denominator and the total number of periods actually attended by the student in all courses, as the numerator. It is desirable for a student to put in 100% attendance in all the subjects. However, a candidate shall be permitted to appear for the semester end examination provided he maintains a minimum of 75% overall attendance in the semester.

The shortage of attendance on medical grounds can be condoned up to a maximum of 9% provided the student puts in at least 66% attendance and provided the Principal is satisfied with the genuineness of the reasons. The Medical Certificates are to be submitted to the Head of the Department when the candidate reports to the classes immediately after the absence. Certificates submitted afterwards shall not be entertained. Condonation fee as fixed by the college for those who put in attendance between $\geq 66\%$ and $<75\%$ shall be charged before the semester-end examinations.

In the case of students who participate in co-curricular, extra-curricular activities like student seminars, N.S.S, N.C.C, Inter-collegiate tournaments and any such other activities involving the representation of the Institute, with the prior approval of the Principal, the candidate may be deemed to have attended the classes during the actual period of such activity, solely for the purpose of attendance.

A student, who could not satisfy the minimum attendance requirement of 66% in any semester, shall be declared 'Detained'. He is not eligible to appear for the semester end examinations. He will not be promoted to the next semester and shall have to repeat that semester with the next batch(es) of students. Such students who are detained and seek readmission, should

submit an undertaking/a declaration that they will abide by the regulations existing at the time of readmission.

X. Minimum Academic Requirements:

The following academic requirements have to be satisfied in addition to the attendance requirements mentioned in item No. IX.

- A student shall be deemed to have satisfied the minimum academic requirements and earned the credits allotted to each theory subject if only he secures not less than 40% marks in the semester-end examination and a minimum of 40% marks in the sum of the internal evaluation and semester-end examination taken together. In the labs/projects, the student should secure a minimum of 50% marks in the external examination and a minimum of 50% marks in the sum of internal evaluation and external examination evaluation taken together.
- Further, a candidate has to secure a minimum of 40 % in theory examination (excluding sessional marks) and a minimum of 50 % (excluding sessional marks) in the Practical Examination / Project / Field Work / Viva Voce / Industrial Training in Semester –End / Year – End Examination and 50% aggregate to pass.
- A student will be promoted to the next semester, if only he satisfies the minimum attendance requirement.
- A student shall be promoted from II Year to III Year only if he fulfills the academic requirement of total 50 % of all credits from regular and supplementary examinations of I Year and II Year – I Semester { i.e., total 3 semesters } examinations, irrespective of whether the candidate takes the examination in all the subjects or not.
- A student shall be promoted from III Year to IV Year only if he fulfills the academic requirements of total 50% of credits from regular and supplementary examinations of I Year, II Year and III Year- I Semester {i.e., total 5 semesters}, irrespective of whether the candidate takes the examinations in all the subjects or not.
- For lateral entry students, there is no credit based restriction for promotion from II year to III year. But a lateral entry student shall be promoted from III year to IV year only if he fulfills the academic requirements of total 50% of credits from regular and supplementary examinations of II year and III year- I Semester {i.e., total 3 semesters} irrespective of whether the candidate takes the examinations in all the subjects or not.

- Students, who fail to complete their B.Tech. Programme within eight academic years from the year of their admission or fail to acquire the credits stipulated for the programme shall forfeit their seat in B.Tech. Programme and their admission shall stand cancelled. For lateral entry students they have to complete the programme in six years from their year of admission.
- A candidate can avail the betterment chances during the validity of all courses.

XI. Award of Grades:

The absolute grading system is adopted as follows:

S.No	Range of marks %	Grade	Grade Points	
1	> 90 ≤ 100	O	10	Out Standing
2	> 80 ≤ 90	A+	9	Excellent
3	> 70 ≤ 80	A	8	Very Good
4	> 60 ≤ 70	B+	7	Good
5	> 55 ≤ 60	B	6	Above Average
6	≥ 50 ≤ 55	C	5	Average
7	≥ 40 < 50	P	4	Pass
8	< 40	F	0	Fail
9			0	Ab (Absent)

Note: Minimum grade to pass in a laboratory course is ‘C’.

The performance of a student at the end of the each semester is indicated in terms of Semester Grade Point Average (SGPA). The SGPA is calculated as below:

$$SGPA = \frac{\sum (\text{Credits of a course} \times \text{Grade points awarded for a course})}{\sum (\text{Credits of a course})}$$

SGPA is calculated for the candidates who have passed in all the courses in that semester.

Cumulative Grade Point Average (CGPA) will be calculated from II semester onwards up to the final semester and its calculation is similar to that of SGPA, considering all the courses offered from the first semester onwards.

CGPA is calculated for those who clear all the courses in all the previous semesters.

XII. Award of Class:

For award of class, a total of best 180 credits are considered in case of four year programme and best 137 credits in case of lateral entry admitted students. A candidate, who becomes eligible for the award of B.Tech.Degree, shall be placed in one of the following classes.

S.No.	Class	CGPA
1	First Class with Distinction	7.0 or more*
2	First Class	6.0 or more but less than 7.0
3	Second Class/Pass	5.0 or more but less than 6.0

***First class with Distinction will be awarded only to those students who clear all the subjects of the program in first attempt of regular examinations.**

The CGPA can be converted to aggregate percentage by multiplying CGPA with 10, in case of requirement by any other university or for any other purpose.

XIII. Eligibility for Award of B.Tech. Degree:

A student shall be eligible for the award of the B.Tech degree if he fulfills all the following conditions:

- 1) Registered and successfully completed all the components prescribed for eligibility in the Programme of study to which he/she is admitted within the stipulated period,
- 2) Obtained CGPA greater than or equal to 5.0 (Minimum requirement for Pass),
- 3) No disciplinary action is pending against him/her and
- 4) Has no dues to the Institute including hostels.

XIV. Malpractices:

The Controller of Examinations/Dean of Examinations shall refer the cases of suspected malpractices in mid examinations and semester-end examinations to Malpractice Enquiry Committee constituted by the Institute. Such committee shall follow the approved scales of punishment. The

Principal shall take necessary action against the erring students based on the recommendations of the committee.

XV. Amendments To Regulations:

The Institute may, from time to time, revise, amend, or change the Regulations, Schemes of Examinations, and / or Syllabi and the changes or amendments made shall be applicable to all the students with effect from the dates notified by the Institute.

XVI. General:

- (i) Where the words ‘he’, ‘him’, ‘his’, occur in the regulations, they include ‘she’, ‘her’, ‘hers’.**
- (ii) The academic regulation should be read as a whole for the purpose of any interpretation.**
- (iii) In case of any doubt or ambiguity in the interpretation of the above rules, the decision of the Principal is final.**

CURRICULUM

First Year I –Semester

Code	Subject name	Instruction periods per week					Max marks		Credits
		Category	Lecture	Tutorial	Practical	Total	Sessional	End marks	
ECE111	English	HS	3	1	-	4	40	60	3
ECE112	Engineering Mathematics I	BS	3	1	-	4	40	60	3
ECE113	Engineering Chemistry	BS	3	1	-	4	40	60	3
ECE114	Professional Ethics & Human Values	HS	2	1	-	3	100	-	2
ECE115	Engineering Physics	BS	3	1	-	4	40	60	3
ECE116	Engineering Chemistry lab	BS	-	-	3	3	50	50	2
ECE117	Programming with C Lab	ES	2	-	3	5	50	50	3
ECEAC1	NCC/ NSS/ Sports	AC	-	-	3	3	-	-	-
Total			16	5	9	30	360	340	19

First Year II –Semester

Code	Subject name	Instruction periods per week					Max marks		Credits
		Category	Lecture	Tutorial	Practical	Total	Sessional	End marks	
ECE121	Engineering Mathematics II	BS	3	1	-	4	40	60	3
ECE122	Applied Physics	BS	3	1	-	4	40	60	3
ECE123	Environmental Sciences	BS	3	1	-	4	40	60	3
ECE124	Engineering Drawing	ES	1	-	3	4	40	60	3
ECE125	Basic Electronics Engineering	ES	3	1	-	4	40	60	3
ECE126	Engineering Physics lab	BS	-	-	3	3	50	50	2
ECE127	Language Lab	HS	-	-	3	3	50	50	2
ECE128	Object Oriented Programming with C++ Lab	ES	2	-	3	5	50	50	3
ECE129	Workshop	ES	-	-	3	3	50	50	2
ECEAC2	NCC/ NSS/ Sports	AC	-	-	3	3	-	-	-
Total			15	4	18	37	400	500	24

BS : Basic Sciences; ES : Engineering Sciences; HS : Humanities and Social Sciences; PC : Professional Core; PE : Professional Elective; OE : Open Elective; PW : Project Work; IT : Industrial Training; AC : Audit Course

Second Year I –Semester

Code	Subject name	Instruction periods per week					Max marks		Credits
		Category	Lecture	Tutorial	Practical	Total	Sessional	End marks	
ECE211	Engineering Mathematics-III	BS	3	1	-	4	40	60	3
ECE212	Electrical Machines	ES	3	1	-	4	40	60	3
ECE213	Data structures	ES	3	1	-	4	40	60	3
ECE214	Signals and Systems	PC	3	1	-	4	40	60	3
ECE215	Network analysis and synthesis	ES	3	1	-	4	40	60	3
ECE216	Electronic Circuits and Analysis-I	PC	4	1	-	5	40	60	4
ECE217	Electronic Circuits and Analysis-I Laboratory	PC	-	-	3	3	50	50	2
ECE218	Network & Electrical Machines Laboratory	ES	-	-	3	3	50	50	2
Total			19	6	6	31	340	460	23

Second Year II –Semester

Code	Subject name	Instruction periods per week					Max marks		Credits
			Lecture	Tutorial	Practical	Total	Sessional	End marks	
ECE221	Engineering Mathematics –IV	BS	3	1	-	4	40	60	3
ECE222	Electronic Circuits and Analysis-II	PC	3	1	-	4	40	60	3
ECE223	Digital Electronics	PC	3	1	-	4	40	60	3
ECE224	Probability Theory & Random Processes	PC	3	1	-	4	40	60	3
ECE225	Electromagnetic Field Theory & Transmission Lines	PC	3	1	-	4	40	60	3
ECE226	Control Systems	ES	3	1	-	4	40	60	3
ECE227	Electronic Circuits and Analysis-II Laboratory	PC	-	-	3	3	50	50	2
ECE228	Simulation Laboratory	PC	-	-	3	3	50	50	2
	Massive Open Online Course (MOOC)	AC	-	-	-	-	-	-	-
Total			18	6	6	30	340	460	22

*MOOCs : Course any time during 2-2 to 4-2. But is grade will be accorded with the 4-2 courses of the program.

Third Year I –Semester

Code	Subject name	Instruction periods per week					Max marks		Credits
		Category	Lecture	Tutorial	Practical	Total	Sessional	End marks	
ECE311	Open Elective- I	OE	3	1	-	4	40	60	3
ECE312	Antennas & Wave Propagation	PC	3	1	-	4	40	60	3
ECE313	Communication Systems Engineering	PC	4	1	-	5	40	60	4
ECE314	Advanced Microprocessors	PC	3	1	-	4	40	60	3
ECE315	Computer Architecture & Organization	ES	3	1	-	4	40	60	3
ECE316	Integrated circuits and Applications	PS	3	1	-	4	40	60	3
ECE317	VHDL Laboratory	PC	-	-	3	3	50	50	2
ECE318	IC Laboratory	PC	-	-	3	3	50	50	2
ECE319	Quantitative Aptitude & Verbal Aptitude-I	HS	4	-	-	4	100	-	2
Total			23	6	6	35	440	460	25

Third Year II –Semester

Code	Subject name	Instruction periods per week					Max marks		Credits
			Lecture	Tutorial	Practical	Total	Sessional	End marks	
ECE321	Microwave Engineering	PC	3	1	-	4	40	60	3
ECE322	Digital Signal Processing	PC	4	1	-	5	40	60	4
ECE323	Microcontrollers & Embedded Systems	PC	3	1	-	4	40	60	3
ECE324	Professional Elective-I	PE	3	1	-	4	40	60	3
ECE325	Digital Communications	PC	3	1	-	4	40	60	3
ECE326	Communication Systems Engineering Laboratory	PC	-	-	3	3	50	50	2
ECE327	Microprocessor & Micro Controllers Lab	PC	-	-	3	3	50	50	2
ECE328	Soft Skills Laboratory	HS	-	-	3	3	100	-	2
ECE 329	Quantitative Aptitude & Verbal Aptitude-II	HS	4	-	-	4	100	-	2
Total			20	5	9	34	500	400	24

Fourth Year I –Semester

Code	Subject name	Instruction periods per week					Max marks		Credits
		Category	Lecture	Tutorial	Practical	Total	Sessional	End marks	
ECE411	Open Elective-II	OE	3	1	-	4	40	60	3
ECE412	Engineering Economics and Management	HS	3	1	-	4	40	60	3
ECE413	Digital Image Processing	PC	3	1	-	4	40	60	3
ECE414	Professional Elective -II	PE	4	1	-	5	40	60	4
ECE415	VLSI Design	PC	3	1	-	4	40	60	3
ECE416	Microwave Engineering Laboratory	PC	-	-	3	3	50	50	2
ECE417	Digital Communications Lab	PC	-	-	3	3	50	50	2
ECE418	Industrial Training Seminar	IT	-	2	2	4	100	-	4
ECE419	Project Phase - I	PW	-	-	8	8	100	-	4
Total			16	7	16	39	500	400	28

Fourth Year II –Semester

Code	Subject name	Instruction periods per week					Max marks		Credits
		Category	Lecture	Tutorial	Practical	Total	Sessional	End marks	
ECE 421	Electronic Measurements and Instrumentation	PC	3	1	-	4	40	60	3
ECE 422	Professional Elective-III	PE	4	1	-	5	40	60	4
ECE 423	Professional Elective-IV	PE	4	1	-	5	40	60	4
ECE 424	Signal and Image Processing Laboratory	PE	-	-	3	3	50	50	2
ECE 425	Project Phase - II & Dissertation	PW	-	-	20	20	100	100	8
ECE 426	Massive Open Online Course (MOOC)	OE	-	-	-	-	100	-	2
Total			11	3	23	37	370	330	23

LIST OF DEPARTMENT ELECTIVES

ELECTIVE SUBJECTS:

Professional Elective-I : ECE 324

1. Analog IC Design
2. EMI / EMC
3. Electronic design and automation theory
4. Telecommunications and switching Networks

Industrial Training during summer vacation after Third Year II - Semester.
But its grade will be accorded with the 4-1 courses of the program.

Professional Elective-II : ECE 414

1. Advanced Digital Signal Processing
2. Radar Signal Processing
3. Satellite Communications & GPS
4. Cellular and Mobile Communications

Professional Elective-III : ECE 422

1. Phased array systems
2. Bio-medical Signal processing
3. VLSI Signal processing
4. Modern Television Engineering

Professional Elective-IV : ECE 423

1. Signal processing algorithms and architecture
2. Design of testability
3. Wireless sensor networks
4. Introduction to Software Defined Radio

OPEN ELECTIVES - I

III Year I - Semester

Department	Name of the Course offered
ELECTRONICS AND COMMUNICATION ENGINEERING	ECE 311 (A) Electronic Design with Integrated Circuits ECE 311 (B) Digital Electronics ECE 311 (C) Applications of Fields and Waves ECE 311 (D) Special Topics: Electronics ECE 311 (E) Applied Electronics
ELECTRICAL AND ELECTRONICS ENGINEERING	EEE 311 Renewable Energy Technologies
MECHANICAL ENGINEERING	MEC 311 (A) Robotics MEC 311 (B) Computer Aided Design
COMPUTER SCIENCE & ENGINEERING	CSE311(A) Computer Operating systems CSE311(B) Fundamentals of Computer Networks CSE311(C) Concepts of Object Oriented Programming CSE311(D) Database Management Systems
INFORMATION TECHNOLOGY	IT 311 (A) Essentials of Information Technology IT 311 (B) Data Structures IT 311 (C) Operating Systems IT 311 (D) Database Management Systems
CHEMICAL ENGINEERING	CHE 311(A) Industrial Safety and Hazards Management CHE 311(B) Engineering Biology CHE 311(C) Fuel Cell Technology CHE 311(D) Design of Experiments
CIVIL ENGINEERING	CIV 311 (A) Basic civil engineering CIV 311 (B) Building Planning and construction CIV 311 (C) Basics of Foundation Engineering
MATHEMATICS	MAT 311 (A) Numerical Methods MAT 311 (B) Fuzzy Set Theory & Fuzzy Logic and its Applications MAT 311 (C) Probability Statistics
PHYSICS	PHY 311 Nano Technology and Engineering Applications
CHEMISTRY	CHY 311 (A) Environmental Sciences CHY 311 (B) Characterisation of Materials

OPEN ELECTIVES - II

IV Year I - Semester

Department	Name of the Course offered
ELECTRONICS AND COMMUNICATION ENGINEERING	ECE 411 (A) Introduction to Embedded System Design ECE 411 (B) Introduction to VLSI Design ECE 411 (C) Introduction to Image Processing /Computer Vision
ELECTRICAL AND ELECTRONICS ENGINEERING	EEE 411 Fundamentals of Electric Power Utilization
MECHANICAL ENGINEERING	MEC 411 (A) Finite Element Analysis MEC 411 (B) Operation research
COMPUTER SCIENCE & ENGINEERING	CSE 411(A) Introduction to soft computing CSE 411(B) Cloud computing overview CSE 411(C) Digital Image processing CSE 411(D) Embedded Systems and Applications
INFORMATION TECHNOLOGY	IT 411 (A) Software Engineering Concepts IT 411 (B) Foundations of Web Development & Design IT 411 (C) Open Source Technologies IT 411 (D) Multimedia Concepts
CHEMICAL ENGINEERING	CHE 411(A) Food Processing Technology CHE 411(B) Corrosion Engineering CHE 411(C) Computational Tools for Engineers CHE 411(D) Bioinformatics
CIVIL ENGINEERING	CIV 411 (A) Elements of Environmental Engineering CIV 411 (B) Water Resources conservation CIV 411 (C) Elements of Transportation Engineering
PHYSICS	PHY 411 Principles & Applications of NDT Methods
CHEMISTRY	CHY 411 (A) Environmental Sciences CHY 411 (B) Green Technologies

FIRST YEAR SYLLABI

**I - Semester
&
II - Semester**

ENGLISH

(Common for all branches)

ECE 111

Credits : 3

Instruction : 3 Periods & 1 Tut/Week

Sessional Marks : 40

End Exam : 3 Hours

End Exam Marks : 60

Course Objectives:

- To improve the language proficiency of the students in English with emphasis on Reading and Writing skills.
- To enable the students to study engineering subjects with greater comprehension & cognizance.
- To strengthen the vocabulary of the students.
- To enable the students to write grammatically correct structures with logical flow.
- To equip the students with the knowledge of different formats of business communication.

Course Outcomes:

By the end of the course, the student will be able to:	
1.	Analyze the structure of the phrases, clauses and sentences
2.	Apply his enriched vocabulary to give better shape to his communication skills.
3.	Effectively use different formats of business correspondence.
4.	Use idiomatic expressions and foreign phrases in his communication.
5.	Analyse, interpret and compose meaningful texts.

SYLLABUS

UNIT I

10 Periods

Vocabulary : One Word Substitutes

Grammar : Noun : Noun Phrase, Gerunds

Writing Skills :

- 1) Formal Letter writing – format, style of letter writing and types of letters — complaint, enquiry, requesting quotations, invitation, regret and acceptance.
- 2) Story Building-Developing a story from the key words, giving a title and describing learning outcomes.

UNIT II

10 Periods

Vocabulary : Foreign phrases or expressions

Grammar : Adjectives : Quantifiers, qualifiers, determiners, nouns as adjectives, verbs as adjectives, adjective phrases

Writing Skills :

1. Technical Report writing – Formal reports and types: Informational reports, Analytical reports and Recommendation reports— Status, feasibility, progress, incident and project.
2. Essay writing.

UNIT III

10 Periods

Vocabulary : Idiomatic expressions- meaning and usage.**Grammar** : Articles (concept and function; definite, indefinite and omission of articles)**Writing Skills :**

1. Preparation of C.V. and Resume-format, style purpose and objective.
2. Précis- writing technique with suitable title.

UNIT IV

9 Periods

Vocabulary : Phrasal Verbs derived from the following dynamic verbs: Go, Get, Run, Take, Look, Put, Hold, Stand etc.**Grammar** : Prepositions or prepositional phrases**Writing Skills:**

1. Reading comprehension – questions based on facts, interpretation, logical deduction, vocabulary.
2. E-mail etiquette- format, style and language

UNIT V

9 Periods

Vocabulary : Synonyms and Antonyms (From the prescribed text only)**Grammar** : Pronouns: Kinds of pronouns, relative pronouns – who and whom, whose, which Verbs - Aspects, moods, tenses, direct and indirect speech (active and passive voice), concord, Infinites and verb participles, verb phrase, Conditionals – probable, improbable, impossible, If-clause, Correction of sentences**TEXT BOOK:***Life through language* Pearson Publication Delhi**REFERENCE BOOKS:**

1. GJ.K. Gangal *A Practical Course for Developing Writing Skill in English* PHI
2. Mark Lester and Larry Beason *Handbook of English Grammar & Usage* Tata McGraw Hill.
3. S.M.Gupta *Current English Grammar And Usage* PHI
4. Dr. P. Prasad, Rajendra K Sharma *The Functional Aspects of Communication Skills* Katson Books
5. AbulHashem *Common errors in English* Ramesh Publishing House
6. M. Ashraf Rizvi *Effective Technical Communication* Tata Mc-Graw Hill
7. Edgar Thorpe & Showick Thorpe *Objective English* Pearson

ENGINEERING MATHEMATICS-I

(Common for all branches)

ECE 112

Credits : 3

Instruction : 3 Periods & 1 Tut/Week

Sessional Marks : 40

End Exam : 3 Hours

End Exam Marks : 60

Course Objectives :

- To impart knowledge in basic concepts of functions of several variables and their applications like maxima & minima.
- To enable the students to study the concepts of Fourier series.
- To enable the students to study the concepts of three dimensional figures like sphere, cone cylinder and conicoids.
- To equip the students with the knowledge of multiple integrals and their applications.
- To introduce the concepts of improper integrals like beta, gamma & error functions.

Course Outcomes:

By the end of the course, student will be able to:	
1.	Familiarize with functions of several variables
2.	Apply Fourier series in solving boundary value problems
3.	Apply the concept of three dimensional analytical geometry
4.	Use mathematical tools needed in evaluating multiple integral and their usage.
5.	Use the concepts of improper integrals, Gamma, Beta and Error functions which are needed in Engineering applications

SYLLABUS

UNIT I

12 Periods

Partial Differentiation: Function of two or more variables – Partial Derivatives – which variable is to be treated as constant – Homogeneous functions – Euler’s theorem – Total Derivative - Change of Variables .Jacobians – Taylor’s theorem for functions of two variables – Maxima and Minima functions of two variables.

UNITII

12 Periods

Fourier series: Introduction – Euler’s formula – conditions for a Fourier expansion – Functions having points of Discontinuity – Change of interval – Even and Odd functions – Half range series-Parseval’s formula.

UNIT III

12 Periods

Three Dimensional Analytical Geometry: Equation of a sphere – Plane section of a sphere – Tangent Plane - Equation of a cone – Right circular cone – Equation of a cylinder – Right circular cylinder.

UNITIV

14 Periods

Multiple Integrals: Double integrals – Change of order of integration – Double integral in polar co-ordinates – Area enclosed by plane curves – Triple Integrals. Volume of Solids- Change of Variables-Area of curved surfaces, Calculation of mass.

UNIT V

10 Periods

Beta & Gamma functions :Beta function – Gamma function relation between Beta and Gamma functions –results and problems, error function.

TEXT BOOK:

1. Dr. B.S. Grewal, Higher *Engineering Mathematics* 43rd edition, Khanna Publishers, New Dehli.

REFERENCE BOOKS:

1. N.P. Bali, Dr . Ashok Saxena, Dr.N.Ch.S. Narayana, *A Text book on Engineering Mathematics* Laxmi pub.(p)Ltd. New Dehli
2. H.K.Dass, *Advanced Engineering Mathematics*, S.chand and company Ltd
3. Dr.M.K. Venkataraman, *Higher Engineering Mathematics* National Pub.Co.Madras.
4. Erwin kreyszig, *Advanced Engineering Mathematics* John Wiley and sons Newyork

ENGINEERING CHEMISTRY

(Common for all branches)

ECE 113

Instruction : 3 Periods & 1 Tut/Week

End Exam : 3 Hours

Credits : 3

Sessional Marks : 40

End Exam Marks : 60

Course Objectives:

- To provide knowledge on problems associated with impure water and various water treatment technologies
- To enable the students to know the importance of semiconducting materials and preparation techniques
- To provide basic knowledge on conventional energy resources, developments in batteries and fuel cells
- To understand the corrosion of metals, various methods to prevent and control of corrosion
- To create awareness on advanced concepts like nano materials, green chemistry and eco-friendly technologies for future development

Course Outcomes:

By end of the course, student will be able to:	
1	Identify the problems associated with raw water in various applications and can adopt suitable technologies for domestic and industrial feed waters.
2	Identify & generalize the properties of semiconducting materials and can select suitable semiconducting & various ceramic materials for specific applications.
3	Classify and analyze the conventional energy sources and design of suitable batteries/cells for different engineering applications.
4	Select and design of suitable materials to prevent corrosion and protect various parts from corrosion.
5	Implement the green chemistry principles, concept of tribology, unique properties of nano & composite materials in designing of suitable methods and materials to meet the technological challenges.
6	Solve scientific problems related to various engineering fields.

SYLLABUS

UNIT I

10 Periods

Water Chemistry: Impurities in water, Hardness of water - units and calcium carbonate equivalents, -estimation of hardness (EDTA method) - disadvantages of hard water, boiler troubles- Scale & Sludge formation - prevention- Internal treatment - (Phosphate, Carbonate and Calgon conditioning) -Caustic embrittlement

Water treatment techniques: Softening of water -lime-soda process -numerical problems on LS process -Zeolite, -ion exchange methods, Desalination of water – Reverse osmosis and Electro dialysis, Municipal water treatment - Screening, sedimentation, coagulation, Sterilization- Chlorination-Break Point chlorination.

UNIT II

10 Periods

Solid State Chemistry: Classification of Solids, Band theory of solids. Chemistry of Semiconductors – Intrinsic, extrinsic, compound and defect semiconductors, Organic semiconductors, Purification and preparation of Semiconductor by zone refining – Single crystal growth(Czochralski method) – epitaxial growth. Liquid crystals, LCD, LED and OLED.

Ceramic Materials: Cement-Manufacture of Portland cement - Setting and hardening of cement -Cement concrete - RCC, Refractories - Classification - properties, Ceramics and its Engineering applications.

UNIT III

10 Periods

Thermal Energy: Fuel –types of fuels -Calorific value and its determination (Bomb calorimeter method) Coal- Ranking of coal - analysis (proximate and ultimate) – COKE – Manufacture (Otto Hoffmann's process). Petroleum – refining of Crude oil; Synthetic petrol – Fisher - Tropsch and Bergius methods, Knocking in Petrol and Diesel engine – Octane number - Cetane number, LPG and CNG.

Chemical Energy: Electrode potential, electro chemical series – Reference electrodes – SHE, Calomel electrode – Galvanic cells – primary cells (Dry cell) secondary cells (Lead acid, Ni-Cd, Li ion batteries) H_2 - O_2 fuel cells.

Solar Energy: Construction and Working of Photovoltaic cell

UNIT IV

08 Periods

Corrosion Chemistry: Origin and theories of corrosion – Types of corrosion - Galvanic corrosion, concentration cell corrosion, pitting corrosion, stress corrosion, inter granular corrosion; Factors affecting corrosion – Corrosion

Prevention & Control of Corrosion:Cathodic protection; Corrosion inhibitors; Protective coatings –Galvanization & Tinning –Anodized coatings - paints & special paints

UNIT V

10 Periods

Nanochemistry: Introduction, growth of nanoparticles (Sol-gel process), Fullerenes and Carbon nanotubes

Green chemistry: Principles of Green chemistry, Alternative Solvents used in green synthesis.

Lubricants: Concept of Tribology -Mechanism of lubrication- Blended oils - properties of lubricating oils -Viscosity Index -Fire & Flash Point -Cloud & Pour Point -Aniline point.

High Polymers & Composites- Basic concepts of Polymers, Effect of polymer structure on properties. Plastics-Thermoplastic and Thermosetting resins, Composites -types- Fiber Reinforced Plastics -Particulate composites -Layer composites, engineering applications of composites.

TEXT BOOK:

1. P.C. Jain and M. Jain *Engineering Chemistry* 16th edition - Dhanapathi Rai & Sons, Delhi

REFERENCE BOOKS:

1. S.S. Dara *A text book of Engineering Chemistry* 15th edition, S. Chand & Co. New Delhi
2. O.G. Palanna *Engineering Chemistry* Tata Mcgraw Hill Education pvt ltd, New Delhi.
3. B.K. Sharma *Engineering Chemistry* - Krishna Prakashan Meerut
4. A.K. Bandopadhyay *Nanomaterials* new age international publishers.
5. V.K. Ahluvalia *Green solvents for organic synthesis* Narosa publications.

PROFESSIONAL ETHICS AND HUMAN VALUES

(Common for All Branches)

ECE114

Credits:2

Instruction: 2 Periods & 1 Tut/Week

Sessional Marks :100

Course Objectives :

- To understand moral values and their significance.
- To draw inspiration for imbibing moral values
- To understand professional ethics and obligations
- To know the code of ethics of relevant Professional societies

Course Outcomes:

By end of the course, student will be able to:	
1.	Understand the right code of conduct.
2.	Assess his/her roles as a proactive member of the society
3.	Solve moral dilemmas and issues
4.	Implement Code of ethics of relevant Professional societies

SYLLABUS

UNIT I

Introduction : Philosophical basis for human values- Human values as enshrined in the Gita, Bible and khoran; Religion- Values propounded in various religions- Need for Religious harmony

UNIT II

Human Values: Inspiration : Inspiration for human values- Mahatma Gandhi, Dr.SarvepalliRadha Krishnan, Swami Vivekananda, Rabindranath Tagore, Mother Theresa- Benefits of Human values- Harmony between Self-interest and human values

UNIT III

Basics of Professional Ethics : Ethical Human Conduct – based on acceptance of basic human values; Humanistic Constitution and Endersal human order – skills, sincerity and fidelity; Scope and characteristics of people-frily and eco-frily production system, Technologies and management systems.

UNIT IV

Professional Ethics in practice : Profession and Professionalism – Professional Accountability, Roles of a professional, Ethics and image of profession; Engineering

Profession and Ethics - Technology and society, Ethical obligations of Engineering professionals, Roles of Engineers in industry, society, nation and the world; Professional Responsibilities – Collegiality, Loyalty, Confidentiality, Conflict of Interest, Whistle Blowing

UNIT V

Indian Constitution, Code of Ethics and Global Issues : Indian Constitution: Fundamental Rights and duties, Freedom, Equality, Fraternity, Justice, Directive principles of state policy. Sample code of Ethics by Professional Societies such as ASME, ASCE, IMEC, IETE, Institution of Engineers (India), Indian Institute of Materials Management etc.

Multinational corporations - Environmental ethics - computer ethics - weapons development - engineers as managers-consulting engineers-engineers as expert witnesses and advisors -moral leadership.

TEXT BOOKS:

1. K.R. Govindan and S.SenthilKumar *Professional Ethics & Human Values* Anuradha Publications.
2. Mike Martin and Roland Schinzinger *Ethics in Engineering* 3rd edition, McGraw Hill. New York (2012).

REFERENCE BOOKS:

1. R. Subramanian *Professional Ethics* Oxford Endersity Press.
2. A.N. Tripathy *Human values* 2003, New Age International Publishers
3. S.B. Srivasthva *Professional Ethics & Human Values* SciTech Publications (India) Pvt. Ltd. New Delhi.
4. Prof. D.R. Kiran *Professional Ethics & Human Values* TATA McGraw Hill Education.
5. M. Govindrajran, S Natrajan & V.S. Senth Kumar *Engineering Ethics (including human Values)* Eastern Economy Edition, Prentice hall of India Ltd

ENGINEERING PHYSICS

(Common for all branches)

ECE 115

Credits : 3

Instruction : 3 Periods & 1 Tut/Week

Sessional Marks : 40

End Exam : 3 Hours

End Exam Marks : 60

Course Objectives:

- To impart knowledge in basic concepts of physics relevant to engineering applications
- To introduce advances in technology for engineering applications

Course Outcomes:

By the end of the course, student will be able to:	
1	Understand the fundamental concepts of thermodynamics.
2	Familiar with the fundamentals of electromagnetic induction and Ultrasonics.
3	Aware of the basic concepts of optics like interference, diffraction, polarization and its various applications.
4	Understand the working principle and applications of lasers and fiber optics.
5	Learn fundamentals of modern physics and its importance in modern technology.

SYLLABUS

UNIT I

10 Periods

Thermodynamics : Heat and work, first law of thermodynamics and its applications, reversible and irreversible processes, heat engine, Carnot cycle and its efficiency, Carnot's theorem, second law of thermodynamics, entropy – entropy change in reversible and irreversible processes, entropy and second law, entropy and disorder, entropy and probability, third law of thermodynamics

UNIT II

10 Periods

Electromagnetism : Faraday's law of induction , Lenz's law, Integral and differential forms of Faraday's law , self-inductance, energy stored in electric and magnetic fields, Poynting vector, displacement current, Maxwell's equations in integral form (no derivation), wave equation, propagation of electromagnetic waves in free space

Ultrasonics: Properties of ultrasonic waves, production of ultrasonic waves by magnetostriction and piezoelectric methods, applications of ultrasonics

UNIT III

10 Periods

Optics

Interference: Introduction, principle of superposition, coherence, Young's double slit experiment, conditions for interference, interference in thin films by reflection, wedge shaped film and Newton's rings

Diffraction: Introduction, Fresnel and Fraunhofer diffraction, diffraction at a single slit

Polarisation: Introduction, types of polarized light, double refraction in uniaxial crystals, Nicol's prism, quarter and half-wave plate, production and detection of plane, circular and elliptically polarized light

UNIT IV

10 Periods

Lasers: Introduction, characteristics of a laser beam, spontaneous and stimulated emission of radiation, population inversion, Ruby laser, He-Ne laser, semiconductor laser, applications of lasers

Fibre optics: Introduction to optical fibers, principle of propagation of light in optical fibers, acceptance angle and acceptance cone, numerical aperture, types of optical fibers, modes of propagation and refractive index profiles, attenuation in optical fibers, advantages of optical fibers in communications, fiber optics communication system, applications of optical fibers, fiber optic sensors

UNIT V

10 Periods

Quantum Mechanics:

Planck's hypothesis, wave-particle duality, introduction to quantum theory, de-Broglie concept of matter waves, Heisenberg's uncertainty principle, Schrodinger's time independent and time dependent wave equations, physical significance and properties of the wave function ψ , application of Schrodinger wave equation for a particle in one dimensional well – eigenwavefunctions and energy eigen values of the particle

Elements of Statistical Mechanics: Elementary concepts of Maxwell-Boltzman, Bose-Einstein and Fermi-Dirac statistics (no derivation)

TEXT BOOKS:

1. S.L Gupta and Sanjeev Gupta *Engineering physics* Dhanpat Rai publications.
2. M.N. Avadhanulu & P.G.Kshirasagar *A text book of engineering physics*, S.Chand publication
3. Resnick & Halliday *Physics* - Volume II

REFERENCE BOOKS:

- 1) V. Rajendran *Engineering physics* McGraw Hill Education Private Ltd
- 2) S.O.Pilai, Sivakami *Engineering Physics* New Age International Publishers
- 3) Young & Freedman *University Physics* Pearson Education
- 4) A.Marikani *Engineering Physics* PHI Learning Private Limited

ENGINEERING CHEMISTRY LAB

(Common for all branches)

ECE 116

Practical / week : 3

End Exam : 3Hrs

Credits : 2

Sessional Marks : 50

End Exam Marks : 50

Course Objectives:

- To provide clear idea over quantitative chemical analysis.
- To improve skills in analyzing samples through titration procedures.
- To familiarize with Instrumental methods of analysis for more accuracy.
- To introduce various methods of analyzing the ore samples.

Course Outcomes:

By end of the course, student will be able to:	
1	Apply experimental skills in quantitative chemical analysis of water quality parameters, substances and ores.
2	Select and use a suitable instrumental technique for the quantitative estimation and analyse the data obtained.

List of Experiments (any 10 experiments are to be completed):

1. Preparation of standard solution
2. Estimation of sodium carbonate present in soda ash.
3. Estimation of amount of calcium present in the Portland cement by titrimetrically.
4. Estimation of amount of Copper present in the Copper ore by Iodometrically.
5. Determination of total Hardness present in the given water sample.
6. Estimation of amount of Zinc by titrating with EDTA.
7. Determine the strength of acid by titrating with strong base using **pH meter**.
8. Estimate the individual strength of acids present in the acid mixture by titrating with strong base using **conductivity meter**.
9. Estimate the amount of Mohr's salt present in the given solution by titrating with potassium dichromate using potentiometer.
10. Determination of viscosity of the given liquid by Ostwald viscometer.

11. Determination of rate constant of acid catalyzed hydrolysis of ester.
12. Determination of partition coefficient of iodine distributed between Water and Carbon tetra chloride.

Demonstration

13. Estimation of amount of dissolved oxygen (D.O) present in the given water sample.
14. Synthesize the Phenol-Formaldehyde resin.

TEXT BOOKS:

1. S.K. Bhasin and SudhaRani *Laboratory manual on Engineering chemistry*, third edition DhanpatRai Publishing Company.

REFERENCE BOOKS:

1. S.S. Dara *Experiments and calculations in Engineering chemistry* 9th edition S. Chand & Company Ltd.

PROGRAMMING WITH C LAB

(Common for all branches)

ECE 117

Credits : 3

Instruction : 2 Periods/Week Practicals : 3 Periods/week Sessional Marks : 50

End Exam : 3 Hrs

End Exam Marks : 50

Course Objectives:

To enable students to

- Understand the program development steps using compilers.
- Strengthen the problem solving skills using programming techniques.
- Design programs using various control structures.
- Develop programs using structures, unions and files.

Course Outcomes:

By the end of the course, student will be able to:	
1.	Gain a working knowledge on programming.
2.	Learn and use the fundamentals of a programming language (such as language-defined data types (int, float, char, double), control constructs (sequence, selection, repetition), program modules (including functions, modules, methods)).
3.	Exhibit the ability to formulate a program that correctly implements the algorithm.
4.	Demonstrate the effective use the programming environment used in the course.

SYLLABUS

1. Overview
2. Introduction to Unix
3. Data Types, Constants
4. Operators, Expressions
5. Control Structures
6. Arrays & Strings
7. Pointers
8. Functions.
9. Structures & Unions
10. Files

REFERENCE BOOKS:

1. YashwantKanetkar*Let Us C* 5th Edition.
2. V.Rajaraman*Fundamentals of Computers* 4th Edition, PHI 2005.
3. Programming Techniques through C, M.G. V. Murthy, Pearson Education, 2002
4. KRVenugopal, SR Prasad *Mastering C* Tata McGraw Hill.
5. B.W. Kernighan, Dennis M. Ritchie *The C – Programming Language* PHI

LIST OF SAMPLE PROGRAMS

1. Write a C program for any three of the following
 - i) To accept the distance between two cities and convert the distance in meters, feet, inches and centimeters. (Note: Input distance in Kilometers).
 - ii) To accept the marks obtained by a student in five different subjects, calculate the total marks and percentage obtained by the student (The maximum marks for each subject is 100).
 - iii) To accept a 3-digit number and calculate the sum of its digits.
 - iv) To accept quantity, product code, unit price of five products and calculate the total price for each product and the SUBTOTAL, TAX, TOTAL and print the details in the following format

Qty	Product code	Unit price	Total price
xx	1	400.00	xxxx.xx
xx	2	20.00	xxxx.xx
xx	3	200.00	xxxx.xx
xx	4	100.00	xxxx.xx
xx	5	200.00	xxxx.xx
		SUB TOTAL	xxxxx.xx
		TAX	xxxx.xx
		TOTAL	xxxxx.xx

- v) To evaluate the following expression
 - a) $(ax + by) / (ax - by)$
 - b) $a^2 + b^2 + \text{squareroot}(2ab)$
2. Write a C program for any three of the following
 - i) To find the maximum and minimum of three numbers.
 - ii) For the above experiment in 1-ii) find and display the grade of the student as prescribed below:

Percentage	Grade
>90	A
>80 and <=90	B
>70 and <=80	C
>60 and <=70	D
>=50 and <=60	E
< 50	F

- iii) To find the roots of a quadratic equation.
- iv) To find the area of a triangle when
 - a) Sides are given
 - b) Base and height are given
 - c) Co-ordinates are given
- v) To accept an alphabet and convert into its opposite case. (Do not use library functions)

3. Write a C program for any four of the following
 - i) To print prime numbers between the specified range (eg. 100 to 200)
 - ii) To generate Pascal triangle format
 - iii) To compute cosine series: $\cos(x) = 1 - x^2/2! + x^4/4! - x^6/6! + \dots$
 - iv) To check whether number is palindrome or not.
 - v) To print set of Armstrong numbers in a specified range. (eg. 100 to 200)
 - vi) To convert the numbers from the following
 - a) Binary to decimal
 - b) Decimal to binary
4. Write a C program to perform the following operations in a given array of 'n' numbers
 - i) Sum of all the numbers
 - ii) Minimum and maximum in the array
 - iii) Searching an element
 - iv) To generate random real numbers in the range of 10 to 20 and sort them.
5. Write a C Program to perform the following on the matrices
 - i) Transpose of a matrix and check the symmetry
 - ii) Trace and norm of a matrix
 - iii) Addition of matrices
 - iv) Multiplication of two matrices
6. Write a C program to perform any two of the following operations on strings (not using library functions)
 - i) To check whether the given string is palindrome or not.
 - ii) To find the length of the string
 - iii) To concatenate two strings.
 - iv) To check whether the given substring exists in a text and display the frequency.
7.
 - i) Write a C program to create a structure for a student with the details name, roll no five subject marks, total marks, percentage and sort the records according to the percentage.
 - ii) Write a C program to add two complex numbers using structures.
 - iii) Write a C program to illustrate difference between union and structure.
8.
 - i) Write a program to calculate the sum of an array using pointers.
 - ii) Write a program to search a name in a given list of names using pointers

9. Write a C program using functions
- i) To illustrate call by value and call by reference
 - ii) To accept a string and character and pass them as parameters to a function, the function shall replace the character in the string with any other specific character and return the modified string.
 - iii) To pass the employee record as a structure to the function. The function shall compute the gross salary (include DA and HRA Calculation), take the savings as input and compute the tax payable as per the prescribed table.

Gross Salary	Tax (%)
Less than 2 Lakhs	NIL
2 Lakhs to 5 Lakhs	10
5 Lakh to 10 Lakh	20
10 Lakhs to 50 Lakhs	30
Above 50 lakhs	50

Note: The employee record shall contain employee name, employee id, hire date, basic salary, DA, HRA.

10. Write a C program for any one program for the following to illustrate recursion
- i) Factorial of a number
 - ii) GCD and LCM of two numbers
 - iii) Fibonacci series
11. Write a C program to perform any three of the following on files
- i) To count the number of alphabets, numbers, words, lines in a given file.
 - ii) To merge two files into third auxiliary file and display the content.
 - iii) To print every even position character in a given file.
 - iv) To separate alphabets and integers into two files from the given source file.
12. Write a C program to update the record of a person in a file by accepting person ID.

Hint:

1. Create the file with few records.
2. The fields in a record
 - a. Name of the person
 - b. Identity (ID) of the person
 - c. Age
 - d. Gender
 - e. Occupation
 - f. Salary

ENGINEERING MATHEMATICS-II

(Common for all branches)

ECE 121

Credits : 3

Instruction : 3 Periods & 1 Tut/Week

Sessional Marks : 40

End Exam : 3 Hours

End Exam Marks : 60

Course Objectives:

- To impart knowledge in basic concepts of solving linear system of equations.
- To enable the students to study the eigen values and eigen vectors of matrix.
- To introduce the concepts of ordinary differential equations and their applications to engineers.
- To enable the students to solve any higher order differential equations and to solve differential equations related to simple electric circuits, Newtons law of cooling.
- To introduce the students to Laplace Transforms and their applications.

Course Outcomes:

By the end of the course, student will be able to:	
1.	Solve linear system equations using of matrix algebra techniques
2.	Determine the Eigen values and vectors of a matrix
3.	Apply different techniques in solving differential equations that model engineering problem
4.	Use the application of Differential equations like simple electric circuits, Newtons law of cooling and to solve any higher order linear ordinary differential equation with constant coefficients
5.	Solve linear differential equations and Network analysis using Laplace transforms.

SYLLABUS

UNIT I

11 Periods

Linear Algebra: Rank of matrix-Elementary Transformation of a matrix- Gauss Jordan Method of finding the inverse – Normal form of the matrix- PAQ form – Consistency of linear system of equations – System of homogeneous and non-homogeneous equations .

UNIT II

12 Periods

Linear transformations – Orthogonal transformations- Vectors (Linearly Independent & Dependent) ,Eigen values , Eigen Vectors, Properties of Eigen values – Cayley Hamilton theorem (without proof).Reduction to diagonal form – Reduction of Quadratic form to canonical form – Nature of quadratic form.,

UNIT III

10 Periods

Differential Equations of first order: First order Linear differential equations , Bernoulli's equations , Exact Differential Equations –Equations reducible to exact Equations - Orthogonal trajectories – Simple Electric circuits-Newton law of cooling.

UNIT IV

10 Periods

Higher order Linear Differential Equations : Definitions – Rules for finding the complementary function, rules for finding the particular integral, method of variation of parameters, equations reducible to linear equations with constant coefficient - Cauchy's homogeneous linear equation, Legendre's linear equation.

UNIT V

17 Periods

Laplace Transforms: Introduction – definitions- Transforms of elementary functions - Properties of Laplace transforms- Transforms of Periodic functions – Transforms of Derivatives – Transforms of Integrals- Multiplication by t^n - division by t -Evaluation of integrals by Laplace transforms.

Inverse Laplace transforms – Other methods of finding inverse transforms (Excluding Residue method) Convolution theorem – Application's to Differential Equations – Unit Step function- Unit Impulsive functions.

TEXT BOOK:

1. Dr. B.S. Grewal *Higher Engineering Mathematics* 43rd edition, Khanna Publishers, New Dehli.

REFERENCE BOOKS:

1. N.P. Bali, Dr . Ashok Saxena, Dr.N.Ch.S. Narayana, *A Text book on Engineering Mathematics* Laxmi pub.(p)Ltd. New Dehli.
2. H.K.Dass, *Advanced Engineering Mathematics*, S.chand and company ltd
3. Dr.M.K. Venkataraman, *Higher Engineering Mathematics* National Pub.Co.Madras.
4. Erwin kreyszig. *Advanced Engineering Mathematics* John Wiley and sons, Newyork.

APPLIED PHYSICS
(for ECE, EEE & Mech)

ECE 122

Credits : 3

Instruction : 3 Periods & 1 Tut/Week

Sessional Marks : 40

End Exam : 3 Hours

End Exam Marks : 60

Course Objectives:

- To enhance student's knowledge of theoretical and modern technological aspects in physics and to introduce fundamentals of physics relevant to engineering applications
- To introduce advances in technology for engineering applications

Course Outcomes:

By end of the course, student will be able to:	
1	Understand the properties of magnetic materials and superconductivity.
2	Understand the dielectric nature of materials, properties and its applications.
3	Aware about nano material properties, synthesis and characterization tools.
4	Familiar with fundamentals of crystal structures.
5	Learn the basic phenomenon involved in semiconductors and semiconductor devices.

SYLLABUS

UNIT I

12 Periods

Magnetic materials: Definition of magnetic permeability, magnetization and magnetic susceptibility, origin of magnetic moment, classification of magnetic materials, properties of diamagnetic and paramagnetic materials, ferromagnetic materials - hysteresis curve, domain theory of ferromagnetism, soft and hard ferromagnetic materials, anti-ferromagnetic and ferrimagnetic materials, ferrites and its applications

Superconductivity: Introduction, properties of superconductors, effect of temperature and magnetic field, Meissner effect, flux quantization, type – I and type – II superconductors, high temperature superconductors, applications of superconductors, BCS theory (qualitative)

UNIT II

10 Periods

Dielectric materials: Definition of electric dipole moment, dielectric polarization and dielectric constant, types of polarization – electronic, ionic and oriental polarization, expression for polarisability, internal fields in solids, Clausius – Mossotti equation, frequency dependence of electronic polarization, properties of ferroelectric materials and their applications

UNIT III

10 Periods

Nanophase materials: Introduction to nanophase materials, properties of nanophase materials, synthesis of nanophase materials – chemical vapour deposition, sol-gel method, MEchanical attrition method, applications of nanophase materials. Principles of X-Ray florescence X-Ray Diffraction- Electron Microscopy (SEM and TEM)

UNIT IV

10 Periods

Crystal structure: Introduction, fundamental terms of crystallography – space lattice, , crystal lattice, unit cell, planes, seven crystal systems – Bravias lattices, cubic lattices, crystal directions and planes, Miller indices, interplanar spacing and interatomic distance , some simple crystal structures, body-centered cubic crystals, face-centered cubic crystals

UNIT V

12 Periods

Semiconductor Physics: Intrinsic and extrinsic semiconductors, Fermi level, carrier concentration in intrinsic semiconductor, continuity equation, direct and indirect band gap semiconductors. Lorentz force, Hall effect and its applications.

Physics of semiconductor devices: open circuited p-n junction diode, energy diagram of p-n diode, working of a diode, volt-ampere characteristics of p-n junction, diode as a rectifier, light emitting diode (LED), liquid crystal display (LCD), photodiode

TEXTBOOKS:

1. S.L Gupta and SanjeevGupta *Engineering physics* DhanpatRai publications.
2. M.N. Avadhanulu&P.G.Kshirasagar *A text book of engineering physics*, S.Chand publication

REFERENCE BOOKS:

1. V.Rajendran *Engineering physics* Tata McGraw Hill Education Private Limited
2. DattuRamanlal Joshi *Engineering Physics* Tata McGraw Hill Education Private Limited
3. A.Marikani *Engineering Physics* PHI Learning Private Limited

ENVIRONMENTAL SCIENCES

(Common for all branches)

ECE 123

Instruction : 3 Periods & 1 Tut/Week

End Exam : 3 Hours

Credits : 3

Sessional Marks : 40

End Exam Marks : 60

Course Objectives:

- To gain knowledge on the importance of environment and ecosystems.
- To acquire knowledge with respect to biodiversity, its threats and its conservation and appreciate the concept of interdependence.
- To acquire knowledge about environmental pollution- sources, effects and control measures of environmental pollution
- To understand the treatment of wastewater and solid waste management.
- To be aware of the national and international concern for environment for protecting the environment

Course Outcomes:

By the end of the course, student will be able to:	
1	Identify the characteristics of various natural resources and can implement the conservation practices
2	Realize the importance of Ecosystem and Biodiversity for maintaining ecological balance
3	Classify, analyze various pollutants and can develop methods for solving problems related to environment
4	Design and evaluate strategies and methods for sustainable development of environmental systems and for the remediation or restoration of degraded environments
5	Get awareness on various environmental laws and regulations applicable to global issues and play a role in solving social problems

SYLLABUS

UNIT I

10 Periods

Introduction to Environment and Natural Resources :

Introduction: Definition, Multidisciplinary nature, Scope and Importance of Environmental Sciences- R & D in environment, green advocacy, green marketing, green media and environment consultancy. Need for public awareness.

Natural Resources: Forest resources-use and overexploitation, deforestation, Big Dams effects on forests and tribal people. Water resources-sources, use and over utilization of surface and ground water, conflicts over water, dams-benefits and problems. Food resources-environmental impact of modern agriculture-fertilizer and pesticides. Land resources-land degradation- landslides, soil erosion and desertification. Energy resources- renewable and non-renewable energy resources and use of alternate-energy sources.

UNIT II

10 Periods

Ecosystem & Bio Diversity

Ecosystem: Concept of an ecosystem-structure and function of an ecosystem Food chains, food webs and ecological pyramids, Energy flow in an ecosystem, Ecosystem regulation, Ecological succession. Types, characteristic features, structure and function of forest, grass land, desert and aquatic ecosystems.

Biodiversity: definition, types, India as a Mega diversity Nation, Values of biodiversity, Hot spots of biodiversity, Threats to biodiversity-habitat loss, poaching, human-wildlife conflicts, Endangered and endemic species, Conservation of biodiversity.

UNIT III

10 Periods

Environmental Pollution And Waste Management : Sources, effects and control measures of Air pollution, Noise Pollution, Soil Pollution, Marine pollution, Thermal pollution, Radio Active Pollution. Water Pollution (Sources, Effects, Control measures, DO, BOD, COD, sewage treatment), Green house effect, Ozone depletion, Acid rain –causes and adverse effects.

Solid waste management: Sources and effects of municipal waste, bio-medical waste, Industrial waste, e-waste, Process of waste management-composting, sanitary landfills, incineration.

UNIT IV

8 Periods

Social Issues And Environment : Social Issues and the Environment: From unsustainable to sustainable development, Environmental Impact Assessment, Water

conservation, Rain water harvesting, water shed management. Resettlement and rehabilitation of people, Environmental ethics.

Urbanization, Industrialization, Transportation, Human population and the environment-population growth, role of information technology in environment and human health.

UNIT V

10 Periods

Legislations, Conventions & Case Studies : Environmental protection act-Air (prevention and control of pollution) act, Water (prevention and control of pollution) act, Wildlife protection act, Forest conservation act.

International Conventions: Stockholm Conference, Brundtland Commission, Rio declaration, Vienna Convention, Kyoto protocol, Johannesburg Summit.

Case Studies: Chipko Movement, Kolleru Lake, Fluorosis, Silent valley project, Narmada BachoAndolan, Ralegaon siddhi, Tehri dam, Madhura refinery and Tajmahal

TEXT BOOK:

1. AnubhaKaushik&C.P.Kaushik*Principles of Environmental Studies* New Age International Publications.

REFERENCE BOOKS:

1. B.K. Sharma *Environmental chemistry*Goel publishing house, Meerut, 2001.
2. G. S. Sodhi*Fundamental concepts of Environmental Chemistry*, Narosa publishing house, New Delhi
3. S .S.Dara*A text book of Environmental Chemistry and pollution control*S.Chand and Company Ltd, New Delhi, 2002.

ENGINEERING DRAWING

(Common for all branches)

ECE 124

Credits : 3

Instruction : 1 Theory & 3 Practical Periods/week

Sessional Marks : 40

End Exam : 3 Hours

End Exam Marks : 60

Course Objectives:

- To increase ability to communicate with people and learn to sketch and take field dimensions.
- To make the student familiar to the drawing practices and convention
- To familiarize the student about various engineering curves used in industry
- To enable the student draft simple engineering components and analyze different views of components.
- To introduce basic Auto CAD skills.

Course Outcomes:

By the end of the course, student will be able to:	
1.	Draw various engineering curves and understand the basic geometrical constructions.
2.	Prepare orthographic projections of points and lines
3.	Produce orthographic projections of plane surfaces
4.	Draw orthographic projections of solids in various orientations.
5.	Prepare isometric projections and understand basics of Computer Aided Drafting.

SYLLABUS

UNIT I

Introduction to Engineering Drawing & basics of geometrical construction. Construction of conic sections, Construction of cycloidal curves (cycloid, epicycloid, and hypocycloid), involutes (over circles and polygon) & Archimedian spiral.

UNIT II

Orthographic projections – projections of points – projections of straight lines (lines parallel to both HP&VP, lines parallel to one and inclined to other, lines inclined to both the planes)

UNIT III

Projections of planes – perpendicular planes – oblique planes

UNIT IV

Projection of solids – Prisms – Cylinder– Pyramids & Cones

UNIT V

Isometric projections – Plane solids, Combination of solids Demonstration & Practice:

Computer aided drafting of lines, planes solids and Dimensioning.

TEXT BOOK:

1. N. D. Bhatt *Engineering Drawing* Charotar Publishing House Pvt. Ltd, 53rd Edition : 2014

REFERENCE BOOKS:

1. K. L. Narayana & P. Kanniah *Engineering Drawing*
2. R. B. Choudary *Engineering Graphics with Auto CAD*
3. Trymbaka Murty *Computer Aided Engineering Drawing*

BASIC ELECTRONICS ENGINEERING

(for ECE branch)

ECE 125

Credits : 3

Instruction : 3 Periods & 1Tut/week

Sessional Marks :40

End Exam : 3Hrs

End Exam Marks : 60

Course Objectives:

- To familiarize the students about different discrete electronic components and CRO.
- To familiarize the students with the analysis and design of Rectifier Circuits.
- To train the students with the operational principle, analysis, design and applications of different types of Diodes.
- To train the students the operational principle, analysis, design and application of different field effect transistors (FET) and circuits using FETs & bipolar junction transistor (BJT).
- To familiarize the students about Analog ICs.

Course Outcomes:

At the end the student will be able to	
1.	Analyze different types of diodes, operation and its characteristics.
2.	Design different types of voltage rectifiers.
3.	Design and analyze the DC bias circuitry of BJT and FET and set up required bias point
4.	Design simple electronic circuits to accomplish a specific function, e.g. DC power supplies, Electronic switches etc.

SYLLABUS

UNITI: Electronic Components

8 periods

Resistors: Types of Resistors- The resistor color code, Variable resistors, Rheostat and Potentiometers, Resistance, Tolerance, Resistivity, Power Ratings of Resistors, Resistor troubles, Ohms Law: Linear proportion between V and I, Choosing a resistor for a circuit, Electric Shock,, Open circuit and Short circuit troubles.

Capacitors: Capacitance, charging and discharging, Typical capacitors, Capacitor Coding, Parallel capacitances, Series capacitances, Energy stored in Electrostatic Field of Capacitance, Measuring and Testing of Capacitors.

Inductors: Self and Mutual Inductance

Semiconductors: Mass Action Law, Mobility, Conductivity, Drift current and Diffusion current, Hall-Effect

UNIT II: Electronic Instruments

8 periods

Types of wire conductors, Connectors, Printed wiring, Switches, Fuses, Wire resistance, Introduction to batteries, Introduction to CRO,CRT, Soldering Materials, Soldering Tools.

UNITIII: Diodes and Applications

8 periods

Semiconductor Materials, The PN Junction Diode, Volt-Amp characteristic curve, Diode approximations, Diode ratings, Rectifier Circuits, Special Diodes.

UNITIV: Transistors

10 periods

Transistor Construction, Transistor Operating region, Transistor Ratings, Transistor Biasing Techniques, Small signal amplify operation, CB,CC,CE configurations, JFET and their Characteristics, Biasing techniques for JFET, MOSFET and their Characteristics, MOSFET Biasing techniques.

UNITV: Integrated Circuits

14 periods

Advantages of ICs over discrete components, Introduction to Op-amp, Differential Amplifiers, Block diagram and Characteristics of Op-Amp, Inverting and Non inverting modes, Virtual ground, CMRR, Slew rate, IC 555 Timer, Block daigram, Modes of operation of IC55, OP-AMP voltage Regulators, Fixed Voltage Regulators (78/79, XX).

TEXT BOOKS:

1. Mitchel E SchultzGrob's *Basic Electronics*, Tata McGraw hill Edition, 10th Edition – (Unit I,II,III,IV)
2. RamaKant A Gayakwad, *Op-Amps and Linear Integrated Circuits*, PHI Fourth Edition-(Unit V)

REFERENCE BOOKS:

1. RG Gupta(2001) *Electronic Instruments and Systems*, Tata McGraw Hill – (Unit II)
2. David A Bell (2008) *Electronic Devices and Circuits*, Oxford University Press. (Unit I,III,IV).

ENGINEERING PHYSICS LAB

(Common for all branches)

ECE 126

Credits : 2

Practical / week : 3

Sessional Marks : 50

End Exam : 3 Hrs

End Exam Marks : 50

Course Objectives:

- To enable the students to acquire skill, technique and utilization of the Instruments

Course Outcomes:

By the end of the course, student will be able to:	
1	Design and conduct experiments as well as to analyze and interpret data.
2	Apply experimental skills to determine the physical quantities related to Heat, Electromagnetism and Optics.

List of experiments (any eight to ten experiments are to be completed)

1. Determination of coefficient of thermal conductivity of a bad conductor- Lee's method.
2. Determination of radius of curvature of a convex lens - Newton's rings.
3. Determination of wavelengths of spectral lines in mercury spectrum-using diffraction grating in normal incidence position.
4. Determination of Cauchy's constants of the material of the prism using spectrometer.
5. Determination of thickness of a thin paper by forming parallel interference fringes-Wedge method.
6. Study of variation of magnetic field along the axis of a current carrying circular coil – Stewart and Gee's apparatus
7. Calibration of a low-range voltmeter using potentiometer.
8. Verification of laws of resistance and determination of specific resistance of wire by using Carey- Foster's bridge.
9. Determination of refractive indices o-ray and e-ray in quartz crystal (double refraction)

10. Determination of the frequency of an electrically maintained tuning fork - Melde's experiment.
11. Determination of Rydberg constant using hydrogen discharge tube.
12. Characteristics of photo cell and determination of Planck's constant – Photoelectric effect.
13. Determination of e/m of an electron by Thomson's method
14. Determination of band gap of semiconductor.

TEXT BOOK:

1. Physics Laboratory Manual prepared by Department of Physics ANITS

REFERENCE BOOKS:

1. D.P Siva Ramaiah and V. Krishna Murthy *Practical physics* Maruti book Depot
2. A.R Vegi *Comprehensive practical Physics* Vegi Publishers Pvt.Ltd.

LANGUAGE LAB
(Common for all branches)

ECE 127

Credits : 2

Practical / week : 3

Sessional Marks : 50

End Exam : 3 Hrs

End Exam Marks : 50

Course Objectives:

- To expose the students to a variety of self-instructional, learner-friendly modes of language learning.
- To facilitate computer-aided multi-media instruction enabling individualized and independent language learning.
- To improve the fluency in spoken English and neutralize mother tongue influence
- To bring about a consistent accent and intelligibility in their pronunciation of English by providing an opportunity for practice in speaking.
- To initiate them into greater use of the computer in resume preparation, report writing, format-making etc.
- To help the students cultivate the habit of reading passages from the computer monitor, thus providing them with the required facility to face computer-based competitive exams such GRE, TOEFL, GMAT etc.

Course Outcomes:

By the end of the course, student will be able to:	
1.	Handle CBT (Computer Based Tests) of the qualifying examinations.
2.	Receive, interpret, remember and evaluate information by practicing effective listening skills.
3.	Speak English with neutralized accent.
4.	Narrate, describe and report incidents and situations using appropriate terminology.

SYLLABUS

I CALL (Computer Aided Language Learning)

1. Introduction to the Sounds of English- Vowels, Diphthongs & Consonants.
2. Introduction to Stress and Intonation.
3. Short and long Reading comprehension exercises (listening skills)
4. Telephoning Skills.

II CSL (Communication Skills Lab)

5. 'Just A Minute' Sessions (JAM).
6. Describing Objects / Situations / People.

7. Video talks
8. Situational Dialogues / Role Play.
9. Oral Presentations- Prepared and Extempore.

Suggested Software

- Cambridge Advanced Learners' English Dictionary with CD.
- English Phonetics and Phonology – 2 CDs set
- English Mastery – Alania ABC
- Telephoning English
- Cambridge Grammar of English (Ronald Carter and Michael McCarthy) CD
- English Grammar in Use -Cambridge University Press
- Communication Skills – Oxford U P (Sanjay Kumar and PushpaLatha)

REFERENCE BOOKS:

Books Suggested for English Language Lab Library (to be located within the lab in addition to the CDs of the text book which are loaded on the systems)

1. *Spoken English (CIEFL)* in 3 volumes with 6 cassettes, OUP.
2. Daniel Jones *English Pronouncing Dictionary* Current Edition with CD.
3. R. K. Bansaland J. B. Harrison, *Spoken English-Orient* Longman 2006 Ed.
4. Dr A Ramakrishna Rao, Dr G Natanam& Prof SA Sankaranarayanan*English Language Communication : A Reader cum Lab Manual*Anuradha Publications, Chennai
5. Krishna Mohan & NP Singh *Speaking English Effectively* (Macmillan)
6. J. Sethi, KamleshSadanand& D.V. Jindal*A Practical Course in English Pronunciation, (with two Audio cassettes)* Prentice-Hall of India Pvt. Ltd., New Delhi.
7. T. Balasubramanian*A text book of English Phonetics for Indian Students* (Macmillan).
8. *English Skills for Technical Students*, WBSCTE with British Council, OL
9. J.K. Gangal *A Practical Course in Effective English Speaking Skills* PHI.

OBJECT ORIENTED PROGRAMMING WITH C++ LAB

(Common for all branches, except for Civil & Chemical branches)

ECE 128

Credits : 3

Instruction : 2 Periods/Week Practicals : 3 Periods/week Sessional Marks : 50

End Exam : 3 Hrs End Exam Marks : 50

Course Objectives :

- To introduce Object Oriented Programming (OOP) using the C++ Language.
- To provide the basic concepts and techniques which form the Object Oriented Programming paradigm.

Course Outcomes:

By the end of the course, student will be able to:	
1.	Understand how to use the programming constructs of CPP.
2.	Use Object Oriented Programming concepts to develop object oriented programs.
3.	Apply various object oriented features to solve real world computing problems using C++ language.

SYLLABUS

List of the experiments to be done on the following topics

1. Overview (Transition from C)
2. OOP Concepts and Characteristics
3. Preprocessor , Command line arguments
4. Classes & Data Abstraction
5. Objects
6. Operator Overloading
7. Inheritance
8. Virtual Functions & Polymorphism
9. I/O Streams
10. Templates
11. File Processing
12. Exception Handling Concepts

REFERENCE BOOKS:

1. Mahesh Bhawe , Sunil patekar *Object Oriented Programming in C++* Second edition , Pearson
2. R Rajaram, *Object Oriented Programming in C++* 2nd Edition New Age International Publishers

3. Herbert Schildt *C++ the Complete Reference* III edition, TMH 1999
4. E Balaguruswamy *Object Oriented Programming with C++* 3rd Edition, McGraw Hill

LIST OF SAMPLE PROGRAMS

1. Write a C++ program that uses a recursive function for solving Towers of Hanoi problem.
2. Write a C++ program to find both the largest and smallest number in a list of integers.
3. Write a C++ program that uses function templates to solve problems 1 and 2 experiments
4. Write a C++ program to implement the matrix ADT using a class. Use operator overloading for implementation
5. Write the definition for a class called **Rectangle** that has floating point data members length and width. The class has the following member functions: **void setlength(float)** to set the length data member **void setwidth(float)** to set the width data member **float perimeter()** to calculate and return the perimeter of the rectangle **float area()** to calculate and return the area of the rectangle **void show()** to display the length and width of the rectangle **int sameArea(Rectangle)** that has one parameter of type Rectangle. sameArea returns 1 if the two Rectangles have the same area, and returns 0 if they don't.
 - i. Write the definitions for each of the above member functions.
 - ii. Write main function to create two rectangle objects. Set the length and width of the first rectangle to 5 and 2.5. Set the length and width of the second rectangle to 5 and 18.9. Display each rectangle and its area and perimeter.
 - iii. Check whether the two Rectangles have the same area and print a message indicating the result. Set the length and width of the first rectangle to 15 and 6.3. Display each Rectangle and its area and perimeter again. Again, check whether the two Rectangles have the same area and print a message indicating the result
6. Create a class called MusicIns to contain three methods string(), wind() and perc(). Each of these methods should initialize string array to contain the following
 - i. Veena, guitar, sitar, sarod and mandolin under string
 - ii. Flute, clarinet, saxophone, nadaswaram and piccolo under wind
 - iii. Table, mridangam, bangos, drums and tambour under percIt should also display the contents of the arrays initialized, create a sub class call TypeIns to contain a method called get() and show().

The get() methods must display a menu as follows

- String instruments
- Wind instruments
- Percussion instruments

The show method should display the relevant details according to user choice the base class variable must be accessible only to its derived classes.

7. Create a base class called shape. It should contain two methods getCoord(), showCoord() to accept x and y co ordinates and to display the same respectively . Create a sub class called Rect. It should contain method to display length and breadth of the rectangle called showCoord() . In main method, execute the showCoord() of Rect class by applying the dynamic method dispatch concept
8. Create a class called car. Initialize the color and body attributes to “blue” and “wagon”. there should be two constructors one is a default the creates blue wagon the other constructor should take two argcolor, body and initialize. write method toString() that returns the color and body. Create a sub class funcar. In sub class there are two constructors to invoke super class constructors resp. Write a method playCD in sub class that displays the message “Beautiful music fills the passenger compartment” execute the methods to show the messages
 - i. Mycar is a blue wagon
 - ii. My father’s car is red convertible.
9. Create the ZooAnimal constructor function. The function has 4 parameters — a character string followed by three integer parameters. In the constructor function dynamically allocate the name field (20 characters), copy the character string parameter into the name field, and then assign the three integer parameters to cageNumber, weightDate, and weight respectively.
10. Write a C++ program to perform operations on complex numbers using operator overloading
11. Write a C++ program to write number 1 to 100 in a data file NOTES.TXT
12. Write a function in C++ to count and display the number of lines not starting with alphabet ‘A’ present in a text file “STORY.TXT”.

Example:
If the file “STORY.TXT” contains the following lines,
The rose is red.
A girl is playing there.
There is a playground.
An aeroplane is in the sky.
Numbers are not allowed in the password.
The function should display the output as 3

WORKSHOP

(Common for all branches)

ECE 129

Credits : 2

Practical / week : 3

Sessional Marks : 50

End Exam : 3 Hrs

End Exam Marks : 50

Course Objectives:

- To provide training and hands on experience to the students on basic Engineering related skills like carpentry, fitting, house wiring and tin smithy.

Course Outcomes:

By the end of the course, student will be able to:	
1.	Make different carpentry joints.
2.	Make simple fitting jobs.
3.	Make simple jobs like funnel, elbow etc. using sheet metal.
4.	Understand and build circuits for different types of applications like stair case wiring, series and parallel connections.

LIST OF EXPERIMENTS

Minimum of three exercises has to be conducted from each trade.

Trade:

Carpentry

1. Cross Lap Joint
2. Dovetail Joint
3. Mortise and Tennon Joint
4. Bridle Joint

Fitting

1. V Fit
2. Square Fit
3. Half Round Fit
4. Dovetail Fit

House Wiring

1. Parallel / Series Connection of three bulbs
2. Stair Case wiring
3. Florescent Lamp Fitting
4. Measurement of Earth Resistance

Tin Smithy

1. Taper Tray
2. Square Box without lid
3. Elbow
4. Funnel